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Chapter 14 1. For the following reaction  $K_c = 0.513$  at 500 K.  $N_2O_4 (g) \rightleftharpoons 2 NO_2 (g)$

If a reaction vessel initially contains an  $N_2O_4$  concentration of 0.0500 M at 500 K, what are the equilibrium concentrations of  $N_2O_4$  and  $NO_2$  at 500 K?

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